

Page	Location	Correction (changes are in bold)
42	Last para, 7 th line	...Lai (1996), Table 2.4-5 (Page 57) ...
58	1 st para.	...valves should be...
64	3 rd para.	...set as high as 1.1 times the vessel MAWP.
71	§2.5.8, 1 st line	Refer to §5.3.4 for...
73	4 th para.	Delete: the nozzle flow model with a discharge coefficient of 0.62 (see §3.6.5.2)
74	2 nd para.	Add: ... options (see §3.6.5.2, page 209)
74	2 nd para.	Delete: ...,either $K_r = 0.1$, or...
81	1 st para.	Delete the last two sentences: "There is ...
81	2 nd para., 3 rd line	... K_R values of full-area rupture disk devices in gas service has been...
81	After 2 nd para.	Insert paragraph P1
81	3 rd para.	Delete: ...for both nozzle ... coefficient is 0.62.
81	3 rd para.	Replace sentence: In the pipe flow model, current certified K_R values (ASME PTC-25, 1994) represent the device flow resistance as that of a full-area flow element with the K_R value included in the total flow resistance of the piping system
81	Last para.	Delete the paragraph
82	1 st para.	Delete the paragraph
82	2 nd para.	Replace paragraph with P2
109	ASME address	The American Society of Mechanical Engineers, 3 Park Avenue , New York, NY 10016-5990
125	3 rd para., last line	...in §3B.4.2.1 .
129	Eq. (3.3-2)	$\beta = ((\rho_{\text{flow}} - \rho_2) / (T_{\text{flow}} - T_2)) / \rho_{\text{flow}}$
129	4 th para.	For a liquid, β can be evaluated from the density change over a 5 °F temperature increment divided by the flowing density (ρ_{flow}).
129	Reference	(1993) API RP 520-I, Appendix C (1997) API RP 521, para. 3.14
131	References	(1993) API RP 520-I, para. 3.3.2 (1997) API RP 521, para. 3.15.1.1 (1993) API RP 520-I, para 3.3.3 (1997) API RP 521, para. 3.15.1.2 (1993) API RP 520-I, para. D.3.2 (1997) API RP 521, para. 3.15.1.2 (1993) API RP 520-I, para. D.5.2.4 (1997) API RP 521, para. 3.15.2.2
133	Table 3.3-2	Vent Rate (SCFH* AIR) Valid at approximately One Atmosphere Pressure
134	References	(1993) API RP 520-I, Table D-2 (1997) API RP 521, Table 4 (1993) API RP 520-I, Table D-3 (1997) API RP 521, Table 5
137	Eq. (3.3-10) and line above	...are approaching zero: (Simpson, 1995a) $W = \frac{C q}{T v_g (dP / dT)_{\text{sat}}} \quad \{(dP / dT)_{\text{sat}} \text{ should be in the denominator}\}$
139	4 th para., 3 rd line	(Simpson, 1995a)
139	6 th para., 5 th line	...component by the following approximation (Simpson, 1995a) :
147	3 rd para., 4 th line	...(page 129)...
148	2 nd para., 5 th line	... Table 3.3-2 ...
148 *	2 nd para., 7 th line	... Eq (3.3-2)
157	1 st para., 3 rd line	...vendors...
161	1 st para., 6 th line	...point is illustrated... {omit be}
166	2 nd line	$\tau = \frac{7998}{335.5} - \left[\frac{160.4 \times 1076}{13.69 \times 0.3451 \times 335.5} \right]^{1/2} = 13.4 \text{ s}$
166	3 rd line	= 3500 lbm {not lbm / s}
178	1 st line	§3B.2.2.5 (Page 269)
178	2 nd para., 2 nd line	§3B.4.2.3 (Page 284)

178	§3.6.1.5, 1 st line	See §3B.3.1.6 (Page 277)
186	Ex., bottom of page	Ell $KF = 0.27$; total $KF = 0.83$; ...from COMFLOW is 66.99 psia {not 67.13}
194	Table note, 3 rd line	$66.82 = 129.9X + (-13.9)(1 - X)$ {delete extra = sign}
197	Table	Liquid Wt %: Acetone – 50, Ethanol – 30, Water – 20
198	Last para., 1 st sentence	Eq. (3B.2-14) (page 258) {not Eq. (3B.2-6) (page 255)}
199	First two lines	The first two lines are duplicated from page 198
199	Eq. (3B.2-18) and (3B.2-19)	Constant should be 0.93028 (3 places)
200	1 st line of Eq. (3B.2.21)	$k = (v / P) [(\partial P / \partial T)_v^2 T / C_v - (\partial P / \partial v)_T]$ {missing – and T should be v }
201	1 st line	Line is a duplicate of the last line on Page 200
201 *	Eq. (3.6-8)	Delete (1 - β') in the denominator
202 *	Definition of β	DELETE
206	2 nd para, 7 th line	Eq. (3.6-8) (page 201)
208	3.6.5.1 Heading	Delete GAS OR VAPOR
209	3.6.5.1	Add following the nomenclature: See 3.6.3 for liquid flow in nozzles
209	3.6.5.2 Heading	Delete GAS OR VAPOR
209	After 1 st para.	Insert paragraph P3
209	2 nd para.	Delete “See 2.6.4 for...”
209	Last para.	Add ...through (r) for the given fluid (K_{RG}, K_{RL}, K_{RGL})
210	1 st para.	Replace paragraph with P4
213	Footnote	Where D is i.d. (inches) of Schedule 40 standard pipe
231	Line after Eq. (3A.3-6)	...data over a limited temperature range (Reid, et al., 1987).
238	Eq. (3A.5-1)	$dv = \left(\frac{\partial v}{\partial T}\right)_p dT + \left(\frac{\partial v}{\partial P}\right)_T dP$ {missing + ; misplaced + }
239	Eq. (3A.5-6)	$\frac{dP}{dT} = \frac{(1 / \rho) (\partial \rho / \partial T)_p + 3\alpha}{(D C_1 / e E) + (1 / \rho) (\partial \rho / \partial P)_T}$ { ρ in place of P in partial derivative}
239 *	Middle of page	The modulus of elasticity is 3×10^7 {not 3×10^6 }
239	Table 3A.5-1	dP/dT , psi / °C (acetic acid...water): 155, 155, 155, 166, 197, 164, 47
246	Eq. (3A.6-6)	ϕ should be in the numerator, not the denominator
256	Eq. (3B.2-8)	$\frac{P_{in}^0}{P_{in}} = \left[\frac{v_{in} G^2 (k - 1)}{2 g_c k P_{in}} + 1 \right]^{k/(k-1)}$ { v_{in} misplaced; no = sign}
258	Reference	(1993) API RP 520-I, para. 4.3.3.1 (2000) API RP 520-I, para. 3.6.3
259	Eq. (3B.2-21)	Last term should be $(dP / dT)_T$
260	References	(1993) API RP 520-I, para. 4.3.3.1 (2000) API RP 520-I, para. 3.6.3 (1993) API RP 520-I, para. 4.3.3.1 (2000) API RP 520-I, para. 3.6.2
262	Eq. (3B.2-29)	$-dP = \frac{1}{g_c} G^2 dv + \left(\frac{dP}{dL}\right)_{fr} dL + \frac{g}{g_c} \frac{1}{v} dZ$ {fr subscript; also in $\left(\frac{dP}{dL}\right)_{fr}$ definition}
265	In place of text starting with “The Churchill values... and ending with Eq. (3B.2-36b)	The friction factor is calculated from the following BASIC-like procedure: $f_1 = 64 / N_{Re}$ If $N_{Re} < 1,000$ Then $f = f_1 / 4$ (laminar f) (3B.2-36) Else $\phi = -2 \log_{10} [(\epsilon / D) / 3.7 + (7 / N_{Re})^{0.9}]$ $f_3 = \{-2 \log_{10} [(\epsilon / D) / 3.7 + 2.51 \phi / N_{Re}]\}^{-2}$

		<p>If $N_{Re} < 10,000$ Then $f_2 = (Re / 13,269)^2$ $f = [f_1^{12} + (f_2^{-8} + f_3^{-8})^{-3/2}]^{1/12} / 4$ (transitional f) (3B.2-36a) Else ($N_{Re} \geq 10,000$) $f = f_3 / 4$ (turbulent f) (3B.2-36b) End If (end of inner “If...Then...Else” statement) End If (end of outer “If...Then...Else” statement)</p>
270	1 st sentence	Add: For incompressible or two-phase flow, equation (3B.2-40) can be solved...
270	1 st para.	Delete: ...and close to unity for gas flow. Add: Do not follow the common practice of using $\omega = 1$ for gas flow.
270	1 st para.	Delete the words: “For any fluid,” Add: “Using physical property information, the value of ...”
270	After 1 st para.	Insert paragraph P5
270	2 nd para., 1 st line	If the flow is choked at P_1 ...
270	3 rd para., 1 st line	Some designers follow the practice of... {delete conservative}
270	Eq. (3B.2-42)	$G_c^2 = (-\partial P / \partial v)_s$ {inverted P and v ; subscript error; sq. rt. (t.)}
273	3 rd para., 1 st line	...program at each... {omit 2 nd the}
273	Table 3B.3-1, Model C	$v / v_A - 1 = a [(P_A / P)^b - 1]$ {missing bracket}
274	3 rd line	◇ TPHEM then sets $b_1 = 1$, $c_0 = 0$, and computes a_0 and b_0 { $c_1 \neq 0$; b_0 , not b_0 }
274	4 th - 6 th line (replace)	For frozen flow use the two-point Model E with $X_B = X_A$ (see §3B.4.3.2.2).
274	1 st para., 3 rd line	§3B.4.2.3
274	§3B.3.1.1, 3 rd line	(somewhat more rigorous for gas flow through pipes).
274	3 rd to last line	Table 3B.3-1 {not Table 3B.23-1}
274	2 nd to last line	$X = 0$
275	Table 3B.3-2	Temperature $T_0 (P_0 / P)^{(k-1)/k}$ T_0 {Error in power}
275	Table 3B.3-2	v_g $v_0 (P / P_0)^{-1/k}$ $v_0 (P / P_0)^{-1}$ {Error in power}
279	Last para., reference	Eq. (3B.2-29)
282	Eq. (3B.4-7)	$G_t^2 = \frac{2 g_c P_0}{v_t^2} \left\{ (1 - X) v_f \left(1 - \frac{P_t}{P_0} \right) + \frac{X v_{g0} k^*}{k^* - 1} \left[1 - \left(\frac{P_t}{P_0} \right)^{(k^*-1)/k^*} \right] \right\}$ {missing brackets and subscript 0 in v_{g0} ; pressure term}
282	Below Eq. (3B.4-7)	$v_t = (1 - X) v_f + X v_g (P_t / P_0)^{-1/k^*}$ {last part should be raised to the power}
283	1 st para.	Eq. (3B.2-18) should be Eq. (3B.2-44)
283	Eq. (3B.4-8)	$g_c G_c^{-2} = \frac{X v_g}{k P_c} + \frac{N_{ne} (v_{fg} / H_{fg})^2 (C_f T - X H_g)}{J}$ {missing - in -2 power}
283	2 nd para., 2 nd to last line	ft ·lb _m / BTU... {not ft ·lb _m / (lbf ·s ²)}
292	Table 3B.4-3	Replace ITPS with IPTS (two places)
292	3 rd line	(Leung, 1995)
292	Table 3B.4-3	For all options, see: “TPHEM – Supplement for Advanced Users” (attached)
294	--	Page is not numbered
294	Table 3B.4-5	For all options, see: “TPHEM – Supplement for Advanced Users” (attached)
294	Table 3B.4-5	Code IC, 3 = Advanced User
295	Table 3b.4-6	Table 3B.4-6
295	Table 3B.4-6	For all options, see: “TPHEM – Supplement for Advanced Users” (attached)
296	Table 3B.4-7	Enthalpy {misspelled}

296 *	Table 3B.4-7	lbm / ft ³	{heading}
296	Table 3B.4-7	Replace Data Set with State	
368	Fig. 5.4-3, 2 nd right box	Read value of y from Figure 5.4-4	{wrong number}
369	2 nd line	...(5.4.10) through (5.4.12)...	
370	Eqs. (5.4.4) and (5.4.5); These equations are preferable.	$D = \sqrt{\frac{4 Q_g (1 - y)}{\pi C u_t (1 - x)}} \quad (5.5.4)$ $C = L / D \text{ (a user-specified length-to-diameter ratio)} \quad (5.5.5)$	
371	3 rd para., 1 st line	Figure 5.4-4	
386	3 rd line from bottom	...sum of partial pressures...	
387	P_l, P_q	Should be partial pressures, except when the condensable and quench liquids are immiscible	
388	• <i>Gas holdup</i>	Replace bullet text with: <ul style="list-style-type: none"> • <i>Gas holdup</i> is the gas trapped in the bubbling liquid. • <i>Entrainment</i>. An allowance for additional gas volume (freeboard) is also needed to minimize entrainment losses. 	
393	Last symbol	H_{q0} = enthalpy of quench liquid at initial temperature	
396 397	P_l, P_q	Should be partial pressures, except when the condensable and quench liquids are immiscible	
397	3 rd line	Equation (5.6.5)...	
403	2 nd para., 4 th line	... partial pressure...	
403	P_{v1}, P_{v2}	= partial pressure of pool...	{not vapor pressure}
434	Eq. (5.9.3)	$M_2 = 1.702 \times 10^{-5} \left(\frac{W}{P D^2} \right) \left(\frac{Z T}{k M_w} \right)^{1/2}$	{missing ½ power}
461	Complete page	May be a duplicate of Pg. 460; see correct page (attached)	
469	2 nd para.	...each component in the vapor leaving...	
487	Reference	API RP 520-I, 7 th Ed. (Jan 2000)	
488	6 th entry	ASME BPVC. Boiler and Pressure Vessel Code, Section VIII, Division 1, Pressure Vessels, 2001 , ASME, New York, NY	
490	Reference	Bluhm, W. C. (1962)	
492	Bottom of page	Some may have poor print quality; see correct page (attached)	
506	Reference	Add: Schmidt, J. and Giesbrecht, H., "Design of Cyclone Separators for Emergency Relief Systems", PSP, 20(1), 6-16 (March 2001)	
508 *	Reference	Straitz (1987a) should be Straitz, J. F., (1977). "Make the Flare Protect the Environment", Hydrocarbon Processing, (56), 131-135.	

* New Addition Since 12/28/04